

# Regiodivergent Enantioselective $\gamma$ -Additions of Oxazolones to 2,3-Butadienoates Catalyzed by Phosphines: Synthesis of $\alpha$ , $\alpha$ -Disubstituted $\alpha$ -Amino Acids and *N*,*O*-Acetal Derivatives

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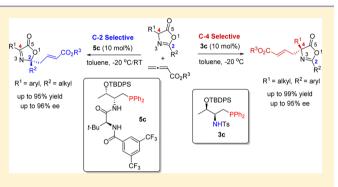
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**Supporting Information** 

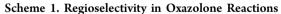
**ABSTRACT:** Phosphine-catalyzed regiodivergent enantioselective C-2- and C-4-selective  $\gamma$ -additions of oxazolones to 2,3butadienoates have been developed. The C-4-selective  $\gamma$ addition of oxazolones occurred in a highly enantioselective manner when 2-aryl-4-alkyloxazol-5-(4*H*)-ones were employed as pronucleophiles. With the employment of 2-alkyl-4aryloxazol-5-(4*H*)-ones as the donor, C-2-selective  $\gamma$ -addition of oxazolones took place in a highly enantioselective manner. The C-4-selective adducts provided rapid access to optically enriched  $\alpha, \alpha$ -disubstituted  $\alpha$ -amino acid derivatives, and the C-2-selective products led to facile synthesis of chiral *N*,*O*-acetals and  $\gamma$ -lactols. Theoretical studies via DFT calculations suggested

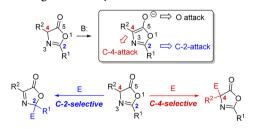


that the origin of the observed regioselectivity was due to the distortion energy that resulted from the interaction between the nucleophilic oxazolide and the electrophilic phosphonium intermediate.

# ■ INTRODUCTION

The chemistry of oxazol-5-(4*H*)-ones (henceforth referred to as oxazolones) has been extensively explored over the past decades, owing to their importance in the synthesis of amino acid derivatives and various heterocyclic structures.<sup>1</sup> Oxazolones have been widely used as a nucleophilic reaction partner due to the relatively high acidity of the  $\alpha$ -proton. As shown in Scheme 1, the enolate intermediate generated upon deproto-





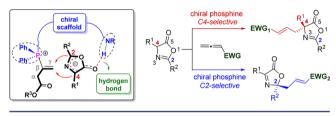
nation of oxazolones may react with electrophiles at different sites, resulting in the formation of different regioisomers. At the outset, we were particularly interested in developing regiodivergent approaches to prepare both C-2- and C-4selective addition products of oxazolones, since the former would allow easy access to disubstituted *N*,*O*-acetals<sup>2</sup> and the latter serve as precursors for the synthesis of  $\alpha$ , $\alpha$ -disubstituted  $\alpha$ -amino acids.<sup>3</sup> A number of transition-metal-mediated reactions of oxazolones were reported, and C-4-selectivity was observed exclusively in all the cases.<sup>4</sup> Recently, organocatalytic approaches employing oxazolones have also been developed. All the existing methods focused on conjugate additions of oxazolones to various activated alkenes, e.g.,  $\alpha_{,\beta}$ -unsaturated carbonyl compounds,<sup>5</sup> nitroolefins,<sup>6</sup> and vinyl sulfones.<sup>7</sup> It is noteworthy that almost all the above additions of oxazolones took place at the C-4-position, except in two examples, whereby C-2-selectivity dominated. While Ooi and co-workers demonstrated that a chiral organic ion pair catalyst could effect a C-2selective addition to  $\alpha,\beta$ -unsaturated acylbenzotriazole,<sup>5b</sup> Jørgensen et al. achieved C-2-selective addition by using acyl phosphonates and chiral thioureas.5d To the best of our knowledge, a versatile strategy enabling highly enantioselective additions of oxazolones at both C-2- and C-4-positions is unknown. Given the importance of both classes of adducts, it is highly desirable to develop a regiodivergent approach for oxazolone addition reactions.

Asymmetric phosphine catalysis has been well-established as a powerful tool for the creation of chiral molecules.<sup>8</sup> Our group has been intensively investigating this research field in recent years. We designed a series of amino acid-based bifunctional

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phosphines and applied them to a wide range of reactions, including (aza)-Morita-Baylis-Hillman reactions,<sup>9</sup> allylic alkylation,<sup>10</sup> Michael addition,<sup>11</sup> and various [3 + 2]/[4 + 2]/[4 + 1] annulation reactions.<sup>12</sup> Phosphine-catalyzed  $\gamma$ -addition reactions<sup>13</sup> are valuable reactions in organic synthesis, and many excellent asymmetric examples emerged in the literature in the past few years.<sup>14</sup> Very recently, we disclosed the utilization of 2,3-butadienoates and prochiral nucleophiles in phosphine-catalyzed  $\gamma$ -addition reactions.<sup>15</sup> Given the acidity of the  $\alpha$ -proton in oxazolone structures, we set out to examine whether it is feasible to utilize oxazolones in phosphinemediated  $\gamma$ -additions.<sup>16a</sup> We envisioned that the phosphonium enolate intermediate should be basic enough to activate oxazolones. The subsequently formed ionic pair between phosphonium and oxazolone-derived enoloate, with the assistance of a hydrogen bonding network, is expected to have a defined structure. We hypothesize that by tuning the chiral bifunctional phosphines, varying oxazolone structures, and employing different allenoates, we may be able to achieve regiodivergent additions of oxazolones to allenoates (Scheme 2). Herein, we document the first example of regiodivergent C-

Scheme 2. Bifuncitonal Phosphine-Catalyzed Enantioselective Regiodivergent  $\gamma$ -Additions of Oxazolones to Allenoates

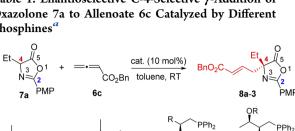


2-selective and C-4-selective  $\gamma$ -additions of oxazolones to allenoates, leading to highly enantioselective preparation of N,O-acetal derivatives and  $\alpha, \alpha$ -disubstituted  $\alpha$ -amino acids, respectively. Moreover, DFT calculations were performed to gain insights into the origin of the observed regioselectivity.

# RESULTS AND DISCUSSION

Phosphine-Catalyzed C-4-Selective and Enantioselective  $\gamma$ -Addition of Oxazolones. Initially, the  $\gamma$ -addition of 2-(4-methoxyphenyl)-4-ethyloxazol-5(4H)-one (7a) to 2,3-butadienoate (6c) was selected as a model reaction, and the catalytic effects of various bifunctional phosphines were evaluated (Table 1). To our delight, all the phosphines examined were effective in promoting the reaction, affording C-4-selective  $\gamma$ -addition adducts as the only product. Bifunctional phosphines with a carbamate, thiourea, or amide were found to be ineffective in asymmetric induction, affording the adducts with low ee values (entries 1-7). Bifunctional phosphines with a sulfonamide group were discovered to be excellent in stereochemical control (entries 8-10), and dipeptide phosphine catalysts were less effective (entries 11-14). In the presence of O-TBDPS-L-threonine-based phosphine sulfonamide 3c, C-4-selective  $\gamma$ -addition product was obtained in high yield and with good enantioselectivity (entry 10).

Having identified the best catalyst 3c, we continued with further optimizations (Table 2). Among all the allenoates examined, the tert-butyl ester (6b) proved to be the best reaction partner (entries 1-8), and toluene remained to be the solvent of choice (entries 9-12). When the reaction was run



Ňн

2a/2b<sup>·</sup> R = Me/i-Pr

OTBDPS

0 ÑН

CF<sub>2</sub>

PPh<sub>2</sub>

2c/2d: R = TBS/TBDPS

OTBDPS

F₂C

Table 1. Enantioselective C-4-Selective  $\gamma$ -Addition of Oxazolone 7a to Allenoate 6c Catalyzed by Different Phosphines<sup>a</sup>

ŇН

1c

NHR

1b: R = Boc

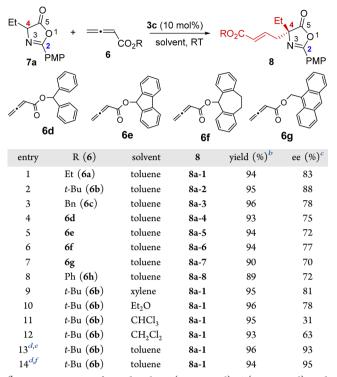
1a: R = COt-Bu

ŃH-	Ts	NHTs		²h₂ OŢĪ́H	-
3a		R = TBS R = TBDPS	t-Bu NHBoc	<i>t</i> -Bu NHR <b>5a/5b</b> : R = Bo <b>5c</b> : R = CO(3,	
entry	catalyst	<i>t</i> (h)	C-4:C-2 <sup>b</sup>	yield (%) <sup>c</sup>	ee (%) <sup>d</sup>
1	1a	15	>20:1	84	11
2	1b	12	>20:1	91	54
3	1c	12	>20:1	93	35
4	2a	12	>20:1	90	41
5	2b	12	>20:1	87	53
6	2c	12	>20:1	91	36
7	2d	12	>20:1	91	38
8	3a	12	>20:1	92	53
9	3b	12	>20:1	95	77
10	3c	12	>20:1	96	81
11	4	18	>20:1	89	19
12	5a	18	>20:1	91	-61
13	5b	18	>20:1	90	-45
14	5c	18	>20:1	89	-37

<sup>a</sup>Reactions were performed with 7a (0.1 mmol), 6c (0.12 mmol), and the catalyst (0.01 mmol) in toluene (1.0 mL) at room temperature. <sup>b</sup>Determined by <sup>1</sup>H NMR analysis of the crude mixture. <sup>c</sup>Isolated yield. <sup>d</sup>Determined by HPLC analysis on a chiral stationary phase. TBDPS = tert-butyldiphenylsilyl, TBS = tert-butyldimethylsilyl, Ts = 4toluenesulfonyl, PMP = 4-methoxyphenyl.

with **6b** in toluene at -20 °C, the desired  $\gamma$ -addition adduct was isolated in 94% yield and with 95% ee (entry 14).

With the optimized reaction conditions for C-4-selective addition of oxazolones in hand, we explored the scope of the reaction (Table 3). Oxazolones with various aliphatic substituents at the C-4-position could be employed, and the C-4-selective adducts were obtained in high yields and with excellent enantioselectivities (entries 1-11). Both linear and branched alkyl groups at the C-4-position were well-tolerated. Remarkably, when tert-butyl-substituted oxazolone was used, enantiomerically enriched  $\gamma$ -addition product was isolated in 91% yield, despite the fact that the newly formed stereogenic center was extremely sterically hindered (entry 7). Notably, the reaction was also well-tolerated for sulfur-containing oxazolones (entries 8 and 9). The presence of a phenyl group in the C-4alkyl-substituted oxazolones slightly lowered the enantioselectivity of the reaction (entries 12 and 13). When oxazolones with different aryl groups at the 2-position were used, the excellent C-4-selectivity and enantioselectivity of the reaction were maintained (entries 14 and 15). However, oxazolones Table 2. Optimization of C-4-Selective  $\gamma$ -Addition Reaction<sup>*a*</sup>



<sup>*a*</sup>Reactions were performed with 7a (0.10 mmol), 6 (0.12 mmol), and 3c (0.01 mmol) in the solvent specified (1.0 mL) at room temperature overnight. <sup>*b*</sup>Isolated yield. <sup>*c*</sup>Determined by HPLC analysis on a chiral stationary phase. <sup>*d*</sup>With 0.15 mmol allenoate. <sup>*e*</sup>The reaction was run at 0 °C for 20 h. <sup>*f*</sup>At -20 °C for 48 h.

with an aryl substitution at the 4-position were found to be unsuitable;<sup>17</sup> although C-4-selective adduct was obtained in excellent yield, the ee value was very low (entry 16). The absolute configurations of the  $\gamma$ -addition products (8) were assigned by comparing the optical rotation of derivative 14 with the value reported in the literature.<sup>18</sup>

Phosphine-Catalyzed C-2-Selective and Enantioselective  $\gamma$ -Addition of Oxazolones. Having established the enantioselective pathway to derive C-4-selective  $\gamma$ -addition products, we next focused on developing  $\gamma$ -addition of oxazolones to allenoates in a C-2-selective fashion. We reasoned that judicious selection of the substrates utilized and careful tuning of the catalyst structures may lead to the discovery of an enantioselective C-2-selective  $\gamma$ -addition. Consequently, different substituted oxazolones and benzyl 2,3-butadienoate were employed, and the results are summarized in Table 4. We were delighted to uncover that replacement of 2-phenyl-4-ethyloxazol-5(4H)-one 7a' with 2methyl-4-phenyloxazol-5(4H)-one 11a completely reverted the regioselectivity, leading to excellent C-2-selective product formation (entries 1 and 2). The above results suggested that the nature of the substituents at the C-2 and C-4 positions of oxazolones seems crucial for the regioselectivity of the  $\gamma$ additions to allenes. Various bifunctional phosphines were subsequently screened, aiming to improve the enantioselectivity of the reaction. While all the phosphines examined afforded C-2-selective adducts, dipeptide phosphines were found to be more effective for asymmetric induction (entries 3-14). When 5c was used, the C-2-selective product N,O-acetal 12a was obtained in 92% yield and with 85% ee (entry 15). The substrate scope for C-2-selective  $\gamma$ -addition of oxazolones to

Table 3. Substrate Scope for 3c-Catalyzed C-4-Selective Enantioselective  $\gamma$ -Addition of Oxazolones 7 to Allenoate  $6b^{a}$ 

R 4	$ \begin{array}{c} 0\\ 5\\ 3\\ 0^{-1}\\ 0^{-2}\\ 7\\ 7 $ Ar	+ =•= CO <sub>2</sub> <i>t</i> -Bu 6b	<b>3c</b> (10 mol%) toluene, -20 °	→ t-BuO <sub>2</sub> C	$ \begin{array}{c}                                     $
en	try	R/Ar	C-4:C-2 <sup>b</sup>	prod./yield (%) <sup>c</sup>	ee (%) <sup>d</sup>
	1	Et/PMP	>20:1	<b>8a</b> /94	95
í	2	Me/PMP	19:1	<b>8b</b> /96	93
3	3	<i>n</i> -Pr/PMP	14:1	8c/99	92
4	4	<i>i</i> -Pr/PMP	>20:1	<b>8d</b> /94	90
:	5	n-Bu/PMP	>20:1	<b>8e</b> /95	91
(	6	I-Bu/PMP	>20:1	<b>8f</b> /94	90
7	7 <sup>e</sup>	t-Bu/PMP	>20:1	<b>8g</b> /91	94
8	8	CH <sub>2</sub> SCH <sub>3</sub> /PMP	>20:1	<b>8h</b> /91	91
9	9	(CH <sub>2</sub> ) <sub>2</sub> SCH <sub>3</sub> /PMP	>20:1	<b>8i</b> /88	93
1	0	<i>n</i> -C <sub>6</sub> H <sub>13</sub> /PMP	>20:1	<b>8</b> j/97	95
1	1	CH(CH <sub>2</sub> ) <sub>5</sub> /PMP	>20:1	<b>8k</b> /96	91
1	2	$(CH_2)_2Ph/PMP$	>20:1	<b>81</b> /95	84
1	3	Bn/PMP	>20:1	<b>8m</b> /93	80
1	4	Et/C <sub>6</sub> H <sub>5</sub>	19:1	<b>8n</b> /94	92
1	5 <sup>f</sup>	$Et/4$ -F- $C_6H_4$	19:1	<b>80</b> /89	91
1	6	Ph/PMP (9)	>20:1	10/89	30

<sup>*a*</sup>Reactions were performed with 7 (0.1 mmol), **6b** (0.15 mmol), and **3c** (0.01 mmol) in toluene (1.0 mL) at -20 °C for 48 h. <sup>*b*</sup>Determined by <sup>1</sup>H NMR of the crude reaction mixture. <sup>*c*</sup>Isolated yield. <sup>*d*</sup>The ee of the major product, determined by HPLC analysis on a chiral stationary phase. <sup>*c*</sup>With 0.02 mmol catalyst **3c**. <sup>*f*</sup>The reaction was stirred for 60 h. PMP = 4-methoxyphenyl.

# Table 4. Enantioselective C-2-Selective $\gamma$ -Addition of Oxazolones: Initial Screenings<sup>*a*</sup>

$\begin{array}{c} & & & & & & & & & & & & & & & & & & &$					
entry	sub.	cat.	C-2:C-4 <sup>b</sup>	prod./yield (%) <sup>c</sup>	ee (%) <sup>d</sup>
1	7a′	3c	<1:20	<b>8a</b> ′/91	74
2	11a	3c	>20:1	<b>12a</b> /92	67
3	11a	1a	7:1	<b>12a</b> /83	16
4	11a	1b	4:1	<b>12a</b> /76	19
5	11a	1c	9:1	<b>12a</b> /86	32
6	11a	2a	10:1	<b>12a</b> /88	46
7	11a	2b	14:1	<b>12a</b> /89	50
8	11a	2c	19:1	<b>12a</b> /90	65
9	11a	2d	>20:1	<b>12a</b> /91	73
10	11a	3a	12:1	<b>12a</b> /90	43
11	11a	3b	19:1	<b>12a</b> /94	51
12	11a	4	>20:1	<b>12a</b> /92	60
13	11a	5a	>20:1	<b>12a</b> /91	-70
14	11a	5b	>20:1	<b>12a</b> /92	-62
15	11a	5c	>20:1	<b>12a</b> /92	-85

<sup>*a*</sup>Reactions were performed with **11a** or **7a**' (0.1 mmol), **6c** (0.12 mmol), and the catalyst (0.01 mmol) in toluene (1.0 mL) at room temperature for 12–18 h. <sup>*b*</sup>Determined by <sup>1</sup>H NMR of the crude reaction mixture. <sup>*c*</sup>Isolated yield. <sup>*d*</sup>The ee of the major product, determined by HPLC analysis on a chiral stationary phase.

allenoates was next evaluated using the optimal conditions identified (Table 5). The reaction worked well for various C-2-

Table 5. Substrate Scope for the C-2-Selective $\gamma$ -Addition <sup><i>a</i></sup>					
$Ar \stackrel{4}{\underset{N \neq 2}{\overset{0}{\underset{1}{\overset{1}{\underset{1}{\overset{1}{\underset{1}{\overset{1}{\underset{1}{\overset{1}{\underset{1}{\underset$					
entry	R/Ar	C-2:C-4 <sup>b</sup>	12/yield $(\%)^c$	ee (%) <sup>d</sup>	
1	Me/Ph	>20:1	12a/92	85	
2	Et/Ph	>20:1	1 <b>2b</b> /90	85	
3 <sup>e</sup>	<i>n</i> -Pr/Ph	>20:1	<b>12c</b> /88	87	
4	<i>i</i> -Pr/Ph	9:1	<b>12d</b> /87	85	
5	<i>n</i> -Bu/Ph	>20:1	12e/93	95	
6	$n-C_5H_{11}/Ph$	12:1	<b>12f</b> /88	91	
7 <sup>f</sup>	C <sub>6</sub> H <sub>11</sub> /Ph	6:1 (19:1)	12g/81 (91)	96 (80)	
8	Et/2-Nap	>20:1	1 <b>2h</b> /95	92	
9 <sup>e</sup>	<i>n</i> -C <sub>5</sub> H <sub>11</sub> /2-Nap	19:1	12i/94	88	
10	Et/1-Nap	>20:1	12j/95	94	
11	<i>n</i> -Pr/1-Nap	>20:1	12k/95	93	
12	Bn/Ph	>20:1	<b>12l</b> /93	80	

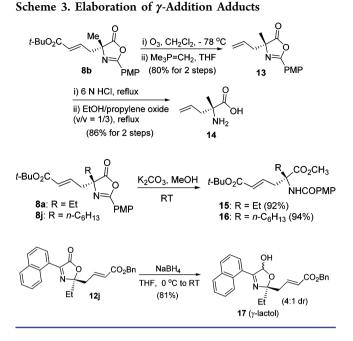
<sup>*a*</sup>Reactions were performed with 11 (0.1 mmol), 6c (0.12 mmol), and the catalyst 5c (0.01 mmol) in toluene (1.0 mL) at room temperature for 12 h. <sup>*b*</sup>Determined by <sup>1</sup>H NMR analysis of the crude mixture. <sup>*c*</sup>Isolated yield. <sup>*d*</sup>The ee of the major product, determined by HPLC analysis on a chiral stationary phase. <sup>*c*</sup>With 0.02 mmol catalyst 5c at 0 °C for 24 h. <sup>*f*</sup>With 0.02 mmol catalyst 5c at -20 °C for 48 h, and the data in parentheses were obtained at room temperature.

substituted oxazolones, although the C-2-selectivities for linear alkyl substituents were superior to those obtained with branched alkyl groups (entries 1-7). Variation of the substituents at the C-4-positions of oxazolones could also be tolerated (entries 8-11). When 2-benzyl-4-phenyl-substituted oxazolone was used, the ee value for the C-2-selective product dropped (entry 12).

Regioselective  $\gamma$ -additions of oxazolones to 2,3-butadienoates offer straightforward synthetic methods to challenging yet valuable molecular architectures. The C-4-selective adducts are not only precursors to  $\alpha, \alpha$ -disubstituted  $\alpha$ -amino acid derivatives but they are also synthetically useful given the rich functionality in the structure.<sup>16</sup> The C-2-selective adducts of this reaction, on the other hand, represent optically enriched N,O-acetals. As illustrated in Scheme 3, adduct 8b could be readily converted to allyl-substituted oxazolone 13 in high vield.<sup>15</sup> Acidic hydrolysis led to ring opening of 13 and simultaneous cleavage of the 4-methoxybenzoyl group, affording  $\alpha, \alpha$ -disubstituted  $\alpha$ -amino acid 14 in high yield. Alternatively, ring opening of oxazolone under mildly basic conditions yielded the corresponding  $\alpha_{,}\alpha_{-}$ disubstituted  $\alpha_{-}$ amino acid derivatives 15 and 16 in excellent yields. When C-2-selective  $\gamma$ -addition product 12j was treated with NaBH<sub>4</sub>, lactol 17 was readily obtained in good yield.

Mechanistic Studies To Understand the Origin of the Observed Regioselectivity. The mechanism of the  $\gamma$ -addition in this report is believed to follow the general mechanism described in the literature,<sup>14,15</sup> and the detailed mechanistic pathways are illustrated in Figure 1. Nucleophilic addition of 3c to 6c yields a zwitterionic intermediate A, which abstracts the C-4-proton of 2-phenyl-4-methyloxazol-5(4H)-one 7a' to form oxazolide C1 and phosphonium B. The subsequent nucleophilic addition takes place at the C-4-

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position of C1 via transition state TS3-C4, leading to the formation of intermediate D1-C4. A hydride shift then takes place and affords intermediate E1-C4, which generates addition product 8' upon elimination of 3c. On the other hand, when 2-methyl-4-phenyloxazol-5(4H)-one 11a is employed, oxazolide C2 is generated. In this pathway, the C-2-selective addition is favored, which eventually leads to the formation of C-2-selective product 12a, via a key transition state TS5-C2.

It was rather striking to discover in this study that the employment of different alkyl or aryl groups at the C-2- or C-4position of oxazolones could result in highly regiodivergent  $\gamma$ additions to allenoates; we thus probed the reaction mechanism<sup>19</sup> by DFT calculations to understand the origin of the observed regioselectivity. The Gibbs free energy profiles of 3c-catalyzed  $\gamma$ -addition reaction of oxazolone 7a' or 11a to allenoate 6c were calculated,<sup>20</sup> and we focused on the step of adding oxazolide (7a'-1/2 or 11a-1/2) to phosphonium (A) to understand the observed regioselectivity (Figure 2). When 2phenyl-4-methyloxazol-5(4H)-one 7a' is used as a substrate, the above addition step can take place via transition state TS3-C2-re at the C-2-position or via transition state TS3-C4-re at the C-4-position. The calculated activation energy for TS3-C4re (C-4-selective) is 2.4 kcal/mol lower than the value for TS3-C2-re (C-2-selective), corresponding to a regioselectivity of 1:56 (C-2:C-4), which is consistent with our experimental observation. Examination of geometries of the two transition states revealed that the bond lengths are comparable, suggesting the steric repulsion is not accountable for the energy difference. To gain more insights, we then applied a distortion/interaction model<sup>21</sup> ( $\Delta E^{\ddagger}_{act} = \Delta E^{\ddagger}_{dist} + \Delta E^{\ddagger}_{int}$ ) and utilized oxazolide and phosphonium as two fragments to further analyze the reaction. The difference of interaction energy terms ( $\Delta E^{\dagger}_{int}$ ) between TS3-C2-re and TS3-C4-re is only 1.4 kcal/mol. However, the difference of distortion energy terms ( $\Delta E^{\dagger}_{dist}$ ) between the two pathways is 3.7 kcal/mol, suggesting that distortion energy played a key role in the observed regioselectivity. When 7a'-1 reacts at the C-2-position with phosphonium to form the new C-C bond, the hybridization state of C-2 changes from sp<sup>2</sup> to sp<sup>3</sup>. In transition state TS3-C2-re, the presence of the 2-phenyl group reduces

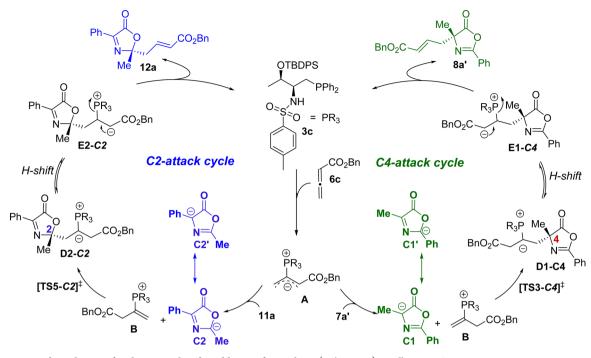


Figure 1. Proposed mechanism for the 3c-catalyzed  $\gamma$ -addition of oxazolone (7a' or 11a) to allenoate 6c.

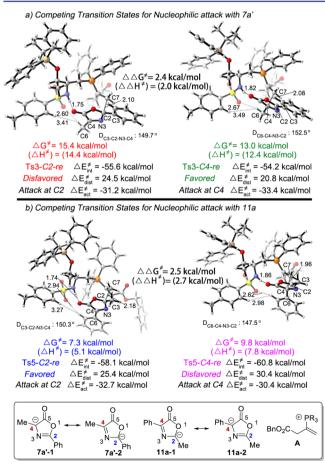


Figure 2. Optimized transition states for 3c-catalyzed nucleophilic attack with 7a' and 11a.

the reactivity of C-2 through conjugation, and it also makes the distortion of the reacting C-2 carbon unfavorable. Furthermore, the dihedral angle of C3–C2–N3–C4 in TS3-C2-re is 149.7°,

2.8° smaller than that in **TS3-***C4-re*, correlating well with the distortion energy difference between the two transition states. Similar analysis was applied to the  $\gamma$ -addition of **11a**. The calculated free energy difference between the two transition states **TS5-C2-re** and **TS5-C4-re** is 2.5 kcal/mol, corresponding to a 1:66 (C-4:C-2) selectivity, which is consistent with the experimental observation. In the less-favored **TS5-***C4-re*, the conjugation of the phenyl group leads to a higher distortion energy. On the other hand, the nonconjugated methyl group at the C-2-position makes the distortion at C-2 easier and accounts for the observed C-2-selectivity of the  $\gamma$ -addition.

The Origin of Observed Enantioselectivity. We also performed DFT calculations to understand the enantioselectivity of the above phosphine-catalyzed  $\gamma$ -addition reactions (Figure 3). The enantioselectivity of the reaction is determined at the nucleophilic attack step. When 7a' is employed, the reface attack occurs through transition state Ts3-C4-re with a barrier of 13.0 kcal/mol, affording intermediate E1-C4-R with an R-configuration. Alternatively, the si-face attack proceeds via transition state Ts3-C4-si with a higher barrier of 16.1 kcal/ mol. The B3LYP-D3 calculations predict a value of 99% ee for the R-isomer, which is consistent with the experimental observation. When substrate 11a is employed, the value of 87% ee predicted by the B3LYP-D3 method based on the energy difference of Ts5-C2-re and Ts5-C2-si is in good agreement with the experimental result. In the geometry of Ts3-C4-si, the H…O distance of 2.52 Å and the H…H distance of 3.06 Å suggest the repulsion between the phenyl group of the reactant and the phosphine catalyst, resulting in a higher transition-state barrier. Similarly, in the geometry of Ts5-C2-si, the short H…H distance of 2.68 Å and the H…C distance of 3.14 Å led to the repulsion between the phenyl group of reactant and the phosphine catalyst, accounting for favorable reface attack.

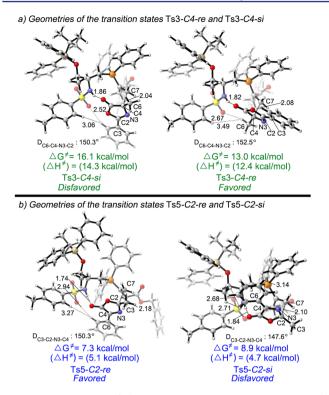


Figure 3. Geometries of the Ts3-C4-re, Ts3-C4-si, Ts5-C2-re, and Ts5-C2-si transition states with 3c as catalyst.

# CONCLUSIONS

In conclusion, we have discovered the first regiodivergent enantioselective  $\gamma$ -additions of oxazolones to 2,3-butadienoates catalyzed by chiral phosphines. By employing 2-aryl-4-alkylsubstituted oxazolones as donors, the C-4-selective  $\gamma$ -addition occurred to furnish highly enantiomerically enriched 4,4disubstituted oxazolones, which are the valuable precursors to  $\alpha_{,\alpha}$ -disubstituted  $\alpha$ -amino acid derivatives. The employment of 2-alkyl-4-aryl- substituted oxazolones as pronucleophiles led to exclusive C-2-selective  $\gamma$ -addition to 2,3-butadienoates, and the adducts are valuable for creation of chiral N,O-acetal and lactols. Disclosed herein is the first practical approach for controlling highly enantioselective C-2- and C-4-selective  $\gamma$ additions of oxazolones to 2,3-butadienoates, leading to facile synthesis of optically enriched  $\alpha_{,\alpha}$ -disubstituted  $\alpha$ -amino acid and  $\gamma$ -lactol derivatives. Our theoretical investigations revealed that the regioselectivity was determined by the distortion energy that resulted from the interactions between the nucleophilic oxazolide and the electrophilic phosphonium intermediate, and the mechanistic insights gained may open up new avenues for the design of regioselective addition processes of oxazolones and other similar donors. Such efforts are currently ongoing in our laboratory and will be reported in due course.

# ASSOCIATED CONTENT

#### **S** Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.5b10524.

Synthetic and experimental details; the characterizations of catalysts, substrates, and products; and the analysis of enantioselectivities of  $\gamma$ -addition adducts (PDF)

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# Notes

The authors declare no competing financial interest.

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### REFERENCES

(1) For reviews, see the following: (a) Fisk, J. S.; Mosey, R. A.; Tepe, J. J. Chem. Soc. Rev. 2007, 36, 1432. (b) Hewlett, N. M.; Hupp, C. D.; Tepe, J. J. Synthesis 2009, 2009, 2825. (c) Mosey, R. A.; Fisk, J. S.; Tepe, J. J. Tetrahedron: Asymmetry 2008, 19, 2755. (d) Alba, A.-N. R.; Rios, R. Chem. - Asian J. 2011, 6, 720.

(2) For an example of racemic synthesis of N,O-acetals, see the following: (a) Harayama, Y.; Yoshida, M.; Kamimura, D.; Wada, Y.; Kita, Y. *Chem. - Eur. J.* **2006**, *12*, 4893 and references cited therein. For examples of asymmetric synthesis from enantiopure starting materials, see the following: (b) Seebach, D.; Aebi, J. D. *Tetrahedron Lett.* **1984**, *25*, 2545. (c) Brunner, M.; Straub, T.; Saarenketo, P.; Rissanen, K.; Koskinen, A. M. P. *Lett. Org. Chem.* **2004**, *1*, 268. For their herbicidal activities, see the following: (d) Ji, Z.; Zhou, F.; Wei, S. Bioorg. Med. Chem. Lett. **2015**, *25*, 4065.

(3) For a review, see: (a) Venkatraman, J.; Shankaramma, S. C.; Balaram, P. Chem. Rev. 2001, 101, 3131. For selected examples, see the following: (b) Cativiela, C.; Díaz-De-Villegas, M. D. Tetrahedron: Asymmetry 2007, 18, 569. (c) Cativiela, C.; Díaz-de-Villegas, M. D. Tetrahedron: Asymmetry 1998, 9, 3517. (d) Ohfune, Y.; Shinada, T. Eur. J. Org. Chem. 2005, 2005, 5127. (e) Tanaka, M. Chem. Pharm. Bull. 2007, 55, 349. (f) Vogt, H.; Bräse, S. Org. Biomol. Chem. 2007, 5, 406.

(4) For selected examples, see the following: (a) Trost, B. M.; Ariza, X. J. Am. Chem. Soc. 1999, 121, 10727. (b) Trost, B. M.; Heinemann, C.; Ariza, X.; Weigand, S. J. Am. Chem. Soc. 1999, 121, 8667. (c) Trost, B. M.; Dogra, K. J. Am. Chem. Soc. 2002, 124, 7256. (d) Trost, B. M.; Jakel, C.; Plietker, B. J. Am. Chem. Soc. 2003, 125, 4438. (e) Trost, B. M.; Czabaniuk, L. C. J. Am. Chem. Soc. 2012, 134, 5778. (f) Melhado, A. D.; Luparia, M.; Toste, F. D. J. Am. Chem. Soc. 2007, 129, 12638. (g) Weber, M.; Jautze, S.; Frey, W.; Peters, R. J. Am. Chem. Soc. 2010, 132, 12222. (h) Chen, W.; Hartwig, J. F. J. Am. Chem. Soc. 2013, 135, 2068.

(5) (a) Cabrera, S.; Reyes, E.; Aleman, J.; Milelli, A.; Kobbelgaard, S.; Jørgensen, K. A. J. Am. Chem. Soc. 2008, 130, 12031. (b) Uraguchi, D.; Ueki, Y.; Ooi, T. Science 2009, 326, 120. (c) Hayashi, Y.; Obi, K.; Ohta, Y.; Okamura, D.; Ishikawa, H. Chem. - Asian J. 2009, 4, 246. (d) Jiang, H.; Paixao, M. W.; Monge, D.; Jørgensen, K. A. J. Am. Chem. Soc. 2010, 132, 2775. (e) Alba, A.-N. R.; Valero, G.; Calbet, T.; Font-Bardía, M.; Moyano, A.; Rios, R. Chem. - Eur. J. 2010, 16, 9884. (f) Alba, A.-N. R.; Valero, G.; Calbet, T.; Font-Bardía, M.; Moyano, A.; Rios, R. New J. Chem. 2012, 36, 613.

(6) (a) Alemán, J.; Milelli, A.; Cabrera, S.; Reyes, E.; Jørgensen, K. A. *Chem. - Eur. J.* **2008**, *14*, 10958. (b) Balaguer, A. N.; Companyo, X.; Calvet, T.; Font-Bardía, M.; Moyano, A.; Rios, R. *Eur. J. Org. Chem.* **2009**, *2009*, 199.

(7) (a) Alba, A.-N. R.; Companyo, X.; Valero, G.; Moyano, A.; Rios, R. Chem. - Eur. J. 2010, 16, 5354. (b) Bravo, N.; Alba, A.-N. R.; Valero, G.; Companyo, X.; Moyano, A.; Rios, R. New J. Chem. 2010, 34, 1816.
(c) Liu, Q.; Qiao, B.; Chin, K. F.; Tan, C.-H.; Jiang, Z. Adv. Synth. Catal. 2014, 356, 3777.

(8) For selected reviews, see the following: (a) Lu, X.; Zhang, C.; Xu, Z. Acc. Chem. Res. **2001**, *34*, 535. (b) Methot, J. L.; Roush, W. R. Adv. Synth. Catal. **2004**, 346, 1035. (c) Ye, L.-W.; Zhou, J.; Tang, Y. Chem.

Soc. Rev. 2008, 37, 1140. (d) Cowen, B. J.; Miller, S. J. Chem. Soc. Rev. 2009, 38, 3102. (e) Marinetti, A.; Voituriez, A. Synlett 2010, 2010, 174. (f) Wang, S.-X.; Han, X.; Zhong, F.; Lu, Y.; Wang, Y. Synlett 2011, 2011 (19), 2766. (g) Zhao, Q.-Y.; Lian, Z.; Wei, Y.; Shi, M. Chem. Commun. 2012, 48, 1724. (h) Wang, Z.; Xu, X.; Kwon, O. Chem. Soc. Rev. 2014, 43, 2927. (i) Wei, Y.; Shi, M. Chem. - Asian J. 2014, 9, 2720. (9) For selected reviews on MBH reactions, see the following: (a) Basavaiah, D.; Reddy, B. S.; Badsara, S. S. Chem. Rev. 2010, 110, 5447. (b) Wei, Y.; Shi, M. Chem. Rev. 2013, 113, 6659. (c) Ma, G.-N.; Jiang, J.-J.; Shi, M.; Wei, Y. Chem. Commun. 2009, 45, 5496. (d) Wei, Y.; Shi, M. Acc. Chem. Res. 2010, 43, 1005. (e) Shi, Y.-L.; Shi, M. Eur. J. Org. Chem. 2007, 2007, 2905. For our examples, see the following: (f) Zhong, F.; Wang, Y.; Han, X.; Huang, K.-W.; Lu, Y. Org. Lett. 2011, 13, 1310. (g) Han, X.; Wang, Y.; Zhong, F.; Lu, Y. Org. Biomol. Chem. 2011, 9, 6734.

(10) Zhong, F.; Luo, J.; Chen, G.-Y.; Dou, X.; Lu, Y. J. Am. Chem. Soc. **2012**, 134, 10222.

(11) Zhong, F.; Dou, X.; Han, X.; Yao, W.; Zhu, Q.; Meng, Y.; Lu, Y. Angew. Chem., Int. Ed. 2013, 52, 943.

(12) For selected examples, see the following: (a) Han, X.; Wang, Y.; Zhong, F.; Lu, Y. J. Am. Chem. Soc. 2011, 133, 1726. (b) Han, X.; Zhong, F.; Wang, Y.; Lu, Y. Angew. Chem., Int. Ed. 2012, 51, 767. (c) Zhong, F.; Han, X.; Wang, Y.; Lu, Y. Chem. Sci. 2012, 3, 1231. (d) Zhong, F.; Han, X.; Wang, Y.; Lu, Y. Angew. Chem., Int. Ed. 2011, 50, 7837. (e) Zhong, F.; Chen, G.-Y.; Han, X.; Yao, W.; Lu, Y. Org. Lett. 2012, 14, 3764. (f) Han, X.; Yao, W.; Wang, T.; Tan, Y. R.; Yan, Z.; Kwiatkowski, J.; Lu, Y. Angew. Chem., Int. Ed. 2014, 53, 5643. (g) Han, X.; Wang, S.-X.; Zhong, F.; Lu, Y. Synthesis 2011, 2011, 1859. (h) Yao, W.; Dou, X.; Lu, Y. J. Am. Chem. Soc. 2015, 137, 54.

(13) For early examples, see the following: (a) Trost, B. M.; Li, C.-J.
J. Am. Chem. Soc. 1994, 116, 10819. (b) Trost, B. M.; Li, C.-J. J. Am.
Chem. Soc. 1994, 116, 3167. (c) Trost, B. M.; Dake, G. R. J. Org. Chem.
1997, 62, 5670. (d) Chen, Z.; Zhu, G.; Jiang, Q.; Xiao, D.; Cao, P.;
Zhang, X. J. Org. Chem. 1998, 63, 5631. (e) Zhang, C.; Lu, X. Synlett
1995, 645.

(14) (a) Chung, Y. K.; Fu, G. C. Angew. Chem., Int. Ed. 2009, 48, 2225. (b) Smith, S. W.; Fu, G. C. J. Am. Chem. Soc. 2009, 131, 14231.
(c) Sinisi, R.; Sun, J.; Fu, G. C. Proc. Natl. Acad. Sci. U. S. A. 2010, 107, 20652. (d) Sun, J.; Fu, G. C. J. Am. Chem. Soc. 2010, 132, 4568.
(e) Fujiwara, Y.; Sun, J.; Fu, G. C. Chem. Sci. 2011, 2, 2196.
(f) Lundgren, R. J.; Wilsily, A.; Marion, N.; Ma, C.; Chung, Y. K.; Fu, G. C. Angew. Chem., Int. Ed. 2013, 52, 2525. (g) Chen, J.; Cai, Y.; Zhao, G. Adv. Synth. Catal. 2014, 356, 359. (h) Fang, Y.-Q.; Tadross, P. M.; Jacobsen, E. N. J. Am. Chem. Soc. 2014, 136, 17966.

(15) (a) Wang, T.; Yao, W.; Zhong, F.; Pang, G. H.; Lu, Y. Angew. Chem., Int. Ed. **2014**, 53, 2964. (b) Wang, T.; Hoon, D. L.; Lu, Y. Chem. Commun. **2015**, 51, 10186. (c) Wang, T.; Yu, Z.; Hoon, D. L.; Huang, K.-W.; Lan, Y.; Lu, Y. Chem. Sci. **2015**, 6, 4912.

(16) For an elegant report by Fu et al. on  $\gamma$ -additions of heterocycles to  $\gamma$ -substituted allenoates, see: (a) Kalek, M.; Fu, G. C. J. Am. Chem. Soc. **2015**, 137, 9438. (b) Wang, D.; Wei, Y.; Shi, M. Chem. Commun. **2012**, 48, 2764. (c) Zou, Y.-Q.; Li, C.; Rong, j.; Yan, H.; Chen, J.-R.; Xiao, W.-J. Synlett **2011**, 2011 (7), 1000.

(17) See the Supporting Information for details; oxazolones with alkyl groups at both C-2 and C-4 positions were too unstable to be prepared.

(18) Lu, T.-J.; Lin, C.-K. J. Org. Chem. 2011, 76, 1621.

(19) (a) Xia, Y.; Liang, Y.; Chen, Y.; Wang, M.; Jiao, L.; Huang, F.; Liu, S.; Li, Y.; Yu, Z.-X. J. Am. Chem. Soc. 2007, 129, 3470. (b) Liang, Y.; Liu, S.; Xia, Y.; Li, Y.; Yu, Z.-X. Chem. - Eur. J. 2008, 14, 4361. (c) Liang, Y.; Liu, S.; Yu, Z.-X. Synlett 2009, 2009, 905. (d) Dudding, T.; Kwon, O.; Mercier, E. Org. Lett. 2006, 8, 3643. (e) Mercier, E.; Fonovic, B.; Henry, C.; Kwon, O.; Dudding, T. Tetrahedron Lett. 2007, 48, 3617. (f) Xie, P.; Lai, W.; Geng, Z.; Huang, Y.; Chen, R. Chem. -Asian J. 2012, 7, 1533. (g) Zhao, L.; Wen, M.; Wang, Z.-X. Eur. J. Org. Chem. 2012, 2012, 3587. (h) Qiao, Y.; Han, K.-L. Org. Biomol. Chem. 2012, 10, 7689. (20) See the Supporting Information for the proposed reaction mechanism (Figure S1) and full Gibbs free energy profiles of the reactions (Figure S2).

(21) For selected examples, see the following: (a) Green, A. G.; Liu,
P.; Merlic, C. A.; Houk, K. N. J. Am. Chem. Soc. 2014, 136, 4575.
(b) Schoenebeck, F.; Ess, D. H.; Jones, G. O.; Houk, K. N. J. Am. Chem. Soc. 2009, 131, 8121. (c) Hayden, A. E.; Houk, K. N. J. Am. Chem. Soc. 2009, 131, 4084. (d) Ess, D. H.; Houk, K. N. J. Am. Chem. Soc. 2007, 129, 10646.